## Preliminary communication

Elucidation of the host-guest geometrical relationship in a branched cyclomaltohexaose inclusion-complex by measurement of the intermolecular nuclear Overhauser effects in the rotating frame

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6-O-( $\alpha$ -D-Glucopyranosyl) cyclomaltohexaose ( $G_1$ - $\alpha$ -CD; Fig. 1) is a branched cyclomaltohexaose ( $\alpha$ -cyclodextrin,  $\alpha$ -CD) that consists of six linked  $(1\rightarrow 4)$ - $\alpha$ -D-glucopyranose residues forming a macrocycle and one branching Dglucopyranose residue attached to one of the residues of the macrocyle through a  $(1\rightarrow 6)$ - $\alpha$ -D-glucopyranosidic linkage. The compound is of interest as it has shown greater water solubility than unsubstituted  $\alpha$ -CDs<sup>1,2</sup>, and, because  $\alpha$ -CDs can accommodate various substances in their cavities by forming inclusion-complexes<sup>3</sup>, it was expected that G<sub>1</sub>-α-CD will also form inclusion-complexes in a manner similar to  $\alpha$ -CD<sup>4</sup>. In order to characterize these CD inclusion-complexes, it was necessary to determine the molecular structure of the conjugates by <sup>1</sup>H-n.m.r. spectroscopy, which has been successfully used to obtain such structural information in solution. Measurement<sup>5</sup> of the host-guest intermolecular <sup>1</sup>H homonuclear Overhauser effect allows the qualitative determination of the orientation of the guest molecule inside of the CD cavity<sup>6,7</sup>. For example, the orientation of the pnitrophenol (pNP) molecule inside the  $\alpha$ -CD cavity has been successfully determined on the basis of a two-dimensional n.O.e. (NOESY) spectrum<sup>7</sup>. While an experiment was attempted on the G<sub>1</sub>-α-CD-pNP complex with experimental parameters similar or identical to those used for the measurement of n.O.e.s of an  $\alpha$ -CD system, no cross-peak due to an intermolecular n.O.e. was observed. Thus it was determined that new experimental parameters must be sought for the NOESY experiment on the  $G_1$ - $\alpha$ -CD system.

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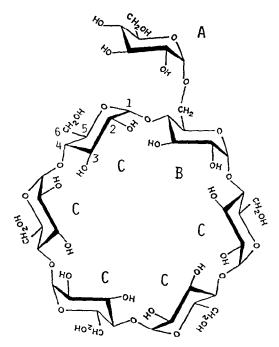


Fig. 1. The structure of  $G_1$ - $\alpha$ -CD. The p-glucopyranosyl residues are divided into three groups, A-C, based on the <sup>1</sup>H-n.m.r. spectrum.

Measurements of n.O.e.s under spin-locked conditions, an operation termed ROESY<sup>8-11</sup>, are particularly useful for molecules with a motional correlation time,  $\tau_c$ , near the condition  $\omega\tau_c=1$ , where  $\omega$  is the angular Larmor frequency. Although the conventional n.O.e. is very small in such cases, an n.O.e. measured in a rotating frame is always positive and quite large. Under conditions used for ROESY experiments, magnetization transfer between scalar coupled spins can also occur<sup>12</sup>, leading to J cross-peaks; however, n.O.e. and J cross-peaks in 2D ROESY spectra can be differentiated by their relative sign with respect to the diagonal peaks<sup>10</sup>. Thus n.O.e. cross-peaks have an opposite phase independent of  $\tau_c$ , while cross-peaks due to J scalar coupling and chemical exchange are in-phase<sup>9,11</sup>, making it possible to distinguish n.O.e. cross-peaks in a 2D spectrum even when complicated J cross-peaks are observed, as in the case of COSY experiments on oligo-saccharides<sup>13</sup>. We now report the results of a rotating-frame n.O.e. study of a  $G_1$ - $\alpha$ -CD inclusion complex with a pNP molecule.

Figure 2 shows the ROESY and the 1D proton spectrum of  $G_1$ - $\alpha$ -CD- and  $\alpha$ -CD-pNP mixtures in  $D_2$ O at pD 7 and 40°. Conditions for effective suppression of coherence transfer from homonuclear Hartmann-Hahn cross polarization was chosen<sup>10</sup>, and assignments of their proton resonances are described elsewhere<sup>7,15</sup>.

In Fig. 2b the contour plot displays only resonances that have opposite phase to the diagonal peaks, i.e., resonances due to transverse n.O.e. effects. A set of

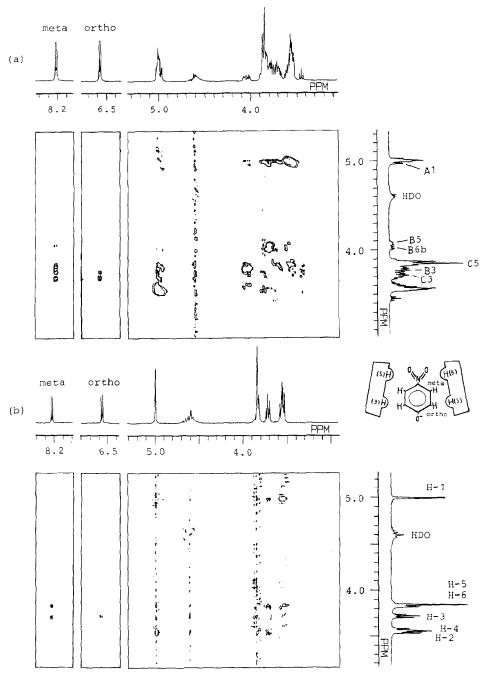


Fig. 2. 500-MHz ROESY spectra with negative contour levels of (a),  $G_1$ - $\alpha$ -CD-pNP at pD 7 and (b),  $\alpha$ -CD-pNP at pD 10 in  $D_2O$ , at 40° with a spin-lock time of 250 ms. The contour plots show the resonances that have a sign opposite to the diagonal peaks. 1D spectra and the signal assignments are also shown, and proton resonances of  $G_1$ - $\alpha$ -CD are expressed as A1, B1, etc., corresponding respectively to the H-1 resonance of the A unit and B unit, etc. In the inset, the spatial relationships for H-3 and H-5 of  $\alpha$ -CD and the *meta* and *ortho* protons of pNP are illustrated.

cross-peaks that connect the H-3 resonance of  $\alpha$ -CD both *meta*- and *ortho*-proton resonances of pNP and connect H-5 to only the *meta* resonance, is observed and essentially identical with the NOESY spectrum previously reported<sup>7</sup>. These cross-peaks due to n.O.e. are subject to no contribution from J cross-peaks, since they arise between two protons belonging to two different molecules. Therefore, they are pure n.O.e. peaks and lead to the same conclusion about the host-guest geometry.

On the other hand, in the  $G_1$ - $\alpha$ -CD-pNP complex, the CD macrocylic ring is made up of at least two types of D-glucopyranose residues: B and C units (see Fig. 1), which exhibit different resonances from each other and consequently make the 1D spectrum complicated. Cross-peaks due to host-guest intermolecular n.O.e.s are observed in Fig. 2a. These include the cross-peaks connecting the H-3 resonances of both the B and C units to both *meta* and *ortho* resonances of pNP and those of the H-5 resonances of both units that connect to only the *meta* resonance of pNP. Therefore, the host-guest geometry in the  $G_1$ - $\alpha$ -CD-pNP complex appears to be similar to that observed in the  $\alpha$ -CD-pNP complex<sup>7</sup>. Consequently, pNP is preferentially inserted into the cavity of  $G_1$ - $\alpha$ -CD with the nitroend first and the *meta*-protons located close to both the H-3 and H-5 of the B and C units so that they exhibit n.O.e. peaks with both protons in each unit.

Similar ROESY experiments should be applicable to other CD inclusion-complexes in solution and should provide experimental data necessary for determining the orientation or insertion depth of a guest molecule into the CD cavity.

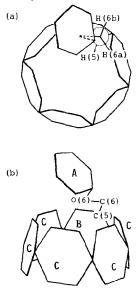


Fig. 3. The molecular structure of  $G_1$ - $\alpha$ -CD expected from n.O.e. data: (a), top view; (b), oblique side view. A hexagon represents one D-glucopyranosyl residue. The C-5-C-6 linkage of the B unit is drawn as the gt rotamer.

Information concerning the conformation of the  $G_1$ - $\alpha$ -CD molecule can be also obtained from the ROESY spectrum. By scrutinizing Fig. 2a, one can determine that some intramolecular n.O.e. cross-peaks are also identified in the region of the CD resonances. The intramolecular n.O.e. cross-peaks (A1,B6a), (B5,B6a), (B6a,B6b), (B4,B6b), and the somewhat weak cross-peak (A1,B6b), taken collectively, suggest a preferential conformation about the C-5–C-6 linkage of the B unit for which the gt rotamer (Fig. 3), where the branch residue A is located above the B unit, is a satisfactory fit. This conformation is consistent with the X-ray structure<sup>14</sup> established on a crystal of  $G_1$ - $\alpha$ -CD ·8H<sub>2</sub>O, which has the gt conformation of the C-5–C-6 linkage of the B unit with the plane comprised of H-1, H-3 and H-5 of the A unit situated nearly parallel with that comprised of the C-1 carbons of the C units. In other words, the branch is not situated over the entrance of the cavity.

The ROESY experiment is a powerful tool for investigating, not only the conformation of molecules in solution undergoing a motion with various time scales, but also for specific interaction between two different molecules.

## **EXPERIMENTAL**

For the n.m.r. sample, solutions, 0.04m in  $G_1$ - $\alpha$ -CD and 0.04m in  $\alpha$ -CD, both containing 0.08m pNP were prepared in  $D_2O$  at pD 7 and pD 10, respectively. According to the reported dissociation constants of the  $G_1$ - $\alpha$ -CD- and  $\alpha$ -CD-pNP complexes<sup>15</sup>, >90% of the  $G_1$ - $\alpha$ -CD and 98% of the  $\alpha$ -CD are considered to be in the complexed state under the experimental conditions used. 2D ROESY spectra were recorded on a JEOL GX-500 spectrometer at 500 MHz and a sweep width of 3500 Hz. The carrier frequency was positioned at 5.58 ppm, and a 4-kHz spin-lock field was used during the mixing period of 250 ms in order to minimize magnetization transfer through scalar couplings<sup>10</sup>. Eight scans were performed for every  $t_1$  value, and two 256 × 2k data sets were acquired. Chemical shifts are given in parts per million (p.p.m.) downfield from that of  $Me_4Si$ .

## REFERENCES

- 1 S. KOBAYASHI, N. SHIBUYA, B. M. YOUNG, AND D. FRENCH, Carbohydr. Res., 126 (1984) 215-224.
- 2 S. KOBAYASHI, K. KAINUMA, AND S. SUZUKI, Agric. Biol. Chem., 51 (1977) 691-698.
- 3 W. SAENGER, Angew. Chem. Int. Ed. Engl., 19 (1980) 344-362.
- 4 Y. Yamamoto, Y. Kanda, Y. Inoue, R. Chújó, and S. Kobayashi, Chem. Lett., (1988) 495–498.
- 5 J. H. NOGGLE AND R. E. SHIRMER, *The Nuclear Overhauser Effect*, Academic Press, New York, 1971.
- 6 R. BERGERON AND R. ROWAN, III, Bioorg. Chem., 5 (1976) 425-436.
- 7 Y. YAMAMOTO, M. ONDA, M. KITAGAWA, Y. INOUE, AND R. CHÚJÓ, *Carbohydr. Res.*, 167 (1987) c11–c16.
- 8 A. A. BOTHNER-BY, R. L. STEPHENS, J. LEE, C. D. WARREN, AND R. W. JEANLOZ, J. Am. Chem. Soc., 106 (1984) 811-813.
- 9 D. G. DAVIS AND A. BAX, J. Magn. Reson., 64 (1985) 533-535.
- 10 A. BAX AND D. G. DAVIS, J. Magn. Reson., 63 (1985) 207-213.

- 11 H. KESSLER, C. GRIESINGER, R. KERSSEBAUM, K. WAGNER, AND R. R. ERNST, J. Am. Chem. Soc., 109 (1987) 607–609.
- 12 L. Braunschweiler and R. R. Ernst, J. Magn. Reson., 53 (1983) 521-528.
- 13 Y. YAMAMOTO, Y. INOUE, R. CHUJO, AND S. KABAYASHI, Carbohydr. Res., 166 (1987) 156-161.
- 14 T. Fujiwara, N. Tanaka, K. Hamada, and S. Kobayashi, Chem. Lett., (1989) 1131-1134.
- 15 Y. KANDA, Y. YAMAMOTO, Y. INOUE, R. CHÜJÖ, AND S. KOBAYASHI, Bull. Chem. Soc. Jpn., 62 (1989) 2002–2008.